

Let's Be The Judge (Elem)

*Gallery walk of questions with incorrect and correct answers.
Determine if right/wrong and why.*

Prep:

1. Chose a concept or standard(s) you want students to review.
2. Select 8-10 examples of questions with correct and incorrect responses..
3. Post each question and selected response for students to examine on chart paper or in area of the room and number them. Another option is to place each question/selected response on a Jamboard page.
4. Place students in groups of 3-4.
5. Provide students with a "Judge's Log" as they go through each question.

Let's Be The Judge

Directions:

1. You will start at a question with your group.
2. Together, read the question and the response given. Discuss if the response is correct or incorrect and justify your answer aloud.
3. Once there is group agreement, record your decision in the numbered box corresponding to the question on your "Judge's Log".
4. When the teacher directs you, your group will move to the next question and repeat the process.

Be prepared to share your Judge's Log with other group members or the whole class at the end of the activity.

Judge's Log

Name: _____ Team Members: _____

#1

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#2

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#3

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#4

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

Judge's Log

Name: _____ Team Members: _____

#5

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#6

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#7

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#8

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

Judge's Log



Name: _____ Team Members: _____

#9

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

#10

Correct or Incorrect: _____

Why or why not?

If incorrect, what is a correct response?

Let's Be The Judge

Example:

- **Concept: Properties of Matter**
- **TEKS: 5.5A** - classify matter based on measurable, testable, and observable physical properties, including mass, magnetism, physical state (solid, liquid, and gas), relative density (sinking and floating using water as a reference point), solubility in water, and the ability to conduct or insulate thermal energy or electric energy. **TEKS: 5.5B** - demonstrate that some mixtures maintain physical properties of their ingredients such as iron filings and sand and sand and water. **TEKS: 5.5C** - identify changes that can occur in the physical properties of the ingredients of solutions such as dissolving salt in water or adding lemon juice to water.
- The next 10 slides are questions/responses.
- The final 10 slides are the answers with TEKS.

Question 1

A student prepared a snack that consisted of grapes, pecans, and strawberries sprinkled with white powdered sugar. The student stored the snack in a refrigerator. An hour later the student observed that the powdered sugar could no longer be seen but the fruit and nuts had not changed in appearance.

What most likely happened to the sugar in the mixture?

Response:

The sugar evaporated at the lower temperature in the refrigerator without causing any changes to the fruit and nuts.

Correct? Incorrect? Why or why not?

Question 2

Students investigate the physical properties of some substances. They draw a table to show how the substances can be grouped. The students need to complete the table with column headings.

Physical Properties of Substances

?	?	?
<ul style="list-style-type: none">• Aluminum foil• Brass key• Gold ring	<ul style="list-style-type: none">• Cooking oil• Soap bubble• Wood chip• Feather	<ul style="list-style-type: none">• Baking soda• Drink mix• White sugar

Which column headings should the students use for their table?

Response:

Good Conductors of Electrical Energy	Less Dense than Water	Soluble in Water
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Correct? Incorrect? Why or why not?

Question 3

A student mixes a sample of stones with a sample of table salt. The mass and volume of the samples were determined before mixing the samples. The mass and volume of each sample is shown.

Material	Grams (g)	Milliliters (mL)
Stones	45	25
Salt	40	35

Response:

The mass of the mixture is 85 grams.

Correct? Incorrect? Why or why not?

Question 4

A student filled each of four beakers with 100 mL of water at 25 °C. The student added an equal amount of a different substance to each of the beakers of water.

Student Investigation

Substance	Appearance	Observations When Stirring	Observations After Stirring Stopped
Iron filings	Silvery gray	Particles swirling around	Particles settled to bottom of beaker
Papain	White powder	Cloudy changing to clear	Clear; no visible particles
Talcum powder	White powder	Floating on surface in clumps	Collected on beaker walls above liquid
Vegetable oil	Yellow liquid	Oil in clumps moving around	Formed a layer on top of water

Based on the student's observations in the table, how many of the substances did NOT dissolve in the water?

Response:

4 substances

Correct? Incorrect? Why or why not?

Question 5

Students conduct an investigation with breakfast cereal. The first four steps of the students' investigation are in the table shown.

Breakfast Cereal Investigation

1. Grind 50 grams of cereal into a fine powder.
2. Stir the cereal powder into 500 milliliters of warm water.
3. Hold a magnet against the side of the beaker at the 250-milliliter mark.
4. Stir the mixture for three minutes.

The students are trying to determine the presence of which substance in the cereal?

Response:

Sugar

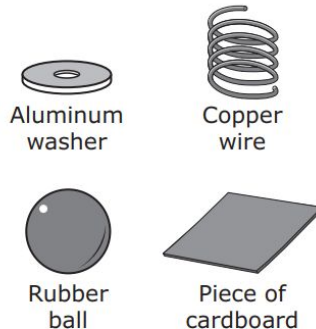
Correct? Incorrect? Why or why not?

Question 6

A student wants to classify four different objects based on physical properties. The student uses the questions shown in the table to test each object.

Materials	Physical Properties		
	Insulate Thermal Energy?	Float in Water?	Conduct Electrical Energy?
1	Yes	No	No
2	No	No	Yes
3	Yes	Yes	No
4	No	No	Yes

Materials



Which statement correctly identifies two of the materials based on the classification of properties in the table?

Response:

Material 2 is an aluminum washer.
Material 3 is a copper wire.

Correct? Incorrect? Why or why not?

Question 7

A student pours 14 grams of sugar into a jar filled with 500 milliliters of water. The student thoroughly stirs the sugar and water to make a solution.

Which change most likely occurs to the sugar when it is added to the water?

Response:

The sugar changes water into a new substance in the solution.

Correct? Incorrect? Why or why not?

Question 8

A table of different properties of four samples of matter is shown.

Sample	Conducts Electricity	Conducts Heat	Soluble in Water	Physical State at Room Temperature
1	No	No	Yes	Solid
2	Yes	Yes	No	Solid
3	No	Yes	Yes	Liquid
4	Yes	Yes	No	Liquid

Which conclusion can be made about the samples based on the table?

Response:

Sample 1 is made of plastic.

Correct? Incorrect? Why or why not?

Question 9

The table lists the ingredients of five different mixtures.

Mixtures and Their Ingredients

Mixture	Ingredients
1	Salt, hot water, sand
2	Sugar, hot water, salt
3	Iron filings and sand
4	Pebbles, wood chips, soil
5	Powdered soap and hot water

In which mixtures do all the ingredients maintain their physical state?

Response:

Mixtures 3 and 4 only

Correct? Incorrect? Why or why not?

Question 10

A student compares the physical properties of the four objects shown.



Rock



Rubber eraser



Paper cup



Wooden block

Which of these physical properties do all four objects have in common?

Response:

They all attract the same metal objects and are not soluble in water.

Correct? Incorrect? Why or why not?

Let's Be The Judge Answer Key

Question 1

A student prepared a snack that consisted of grapes, pecans, and strawberries sprinkled with white powdered sugar. The student stored the snack in a refrigerator. An hour later the student observed that the powdered sugar could no longer be seen but the fruit and nuts had not changed in appearance.

What most likely happened to the sugar in the mixture?

Response:

The sugar evaporated at the lower temperature in the refrigerator without causing any changes to the fruit and nuts.

Correct Response:

The sugar dissolved in the moisture on the fruit.

TEKS: 5.5C - identify changes that can occur in the physical properties of the ingredients of solutions such as dissolving salt in water or adding lemon juice to water

Question 2

Students investigate the physical properties of some substances. They draw a table to show how the substances can be grouped. The students need to complete the table with column headings.

Physical Properties of Substances

?	?	?
<ul style="list-style-type: none">• Aluminum foil• Brass key• Gold ring	<ul style="list-style-type: none">• Cooking oil• Soap bubble• Wood chip• Feather	<ul style="list-style-type: none">• Baking soda• Drink mix• White sugar

Which column headings should the students use for their table?

Response:

Good Conductors of Electrical Energy	Less Dense than Water	Soluble in Water
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Correct Response:

What is listed above is correct.

TEKS: 5.5A - classify matter based on measurable, testable, and observable physical properties, including mass, magnetism, physical state (solid, liquid, and gas), relative density (sinking and floating using water as a reference point), solubility in water, and the ability to conduct or insulate thermal energy or electric energy

Question 3

A student mixes a sample of stones with a sample of table salt. The mass and volume of the samples were determined before mixing the samples. The mass and volume of each sample is shown.

Material	Grams (g)	Milliliters (mL)
Stones	45	25
Salt	40	35

Response:

The mass of the mixture is 85 grams.

Correct Response:

The mass of the mixture is 85 grams.

TEKS: 5.5B - demonstrate that some mixtures maintain physical properties of their ingredients such as iron filings and sand and sand and water

Question 4

A student filled each of four beakers with 100 mL of water at 25 °C. The student added an equal amount of a different substance to each of the beakers of water.

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Vegetable oil	Yellow liquid	Oil in clumps moving around	Formed a layer on top of water

Based on the student's observations in the table, how many of the substances did NOT dissolve in the water?

Response:

4 substances

Correct Response:

3 substances

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3. Hold a magnet against the side of the beaker at the 250-milliliter mark.
4. Stir the mixture for three minutes.

The students are trying to determine the presence of which substance in the cereal?

Response:

Sugar

Correct Response:

Iron

TEKS: 5.5B - demonstrate that some mixtures maintain physical properties of their ingredients such as iron filings and sand and sand and water

Question 6

A student wants to classify four different objects based on physical properties. The student uses the questions shown in the table to test each object.

Materials	Physical Properties		
	Insulate Thermal Energy?	Float in Water?	Conduct Electrical Energy?
1	Yes	No	No
2	No	No	Yes
3	Yes	Yes	No
4	No	No	Yes

Materials



Aluminum washer



Copper wire



Rubber ball



Piece of cardboard

Which statement correctly identifies two of the materials based on the classification of properties in the table?

Response:

Material 2 is an aluminum washer. Material 3 is a copper wire.

Correct Response:

Material 3 is a piece of cardboard. Material 4 is an aluminum washer.

TEKS: 5.5A - classify matter based on measurable, testable, and observable physical properties, including mass, magnetism, physical state (solid, liquid, and gas), relative density (sinking and floating using water as a reference point), solubility in water, and the ability to conduct or insulate thermal energy or electric energy

Question 7

A student pours 14 grams of sugar into a jar filled with 500 milliliters of water. The student thoroughly stirs the sugar and water to make a solution.

Which change most likely occurs to the sugar when it is added to the water?

Response:

The sugar changes water into a new substance in the solution.

Correct Response:

The sugar completely dissolves in the solution.

TEKS: 5.5C - identify changes that can occur in the physical properties of the ingredients of solutions such as dissolving salt in water or adding lemon juice to water

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3	No	Yes	Yes	Liquid
4	Yes	Yes	No	Liquid

Which conclusion can be made about the samples based on the table?

Response:

Sample 1 is made of plastic.

Correct Response:

Sample 2 is made of metal.

TEKS: 5.5A - classify matter based on measurable, testable, and observable physical properties, including mass, magnetism, physical state (solid, liquid, and gas), relative density (sinking and floating using water as a reference point), solubility in water, and the ability to conduct or insulate thermal energy or electric energy

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In which mixtures do all the ingredients maintain their physical state?

Response:

Mixtures 3 and 4 only

Correct Response:

Mixtures 3 and 4 only

TEKS: 5.5B - demonstrate that some mixtures maintain physical properties of their ingredients such as iron filings and sand and sand and water

Question 10

A student compares the physical properties of the four objects shown.



Rock



Rubber eraser



Paper cup



Wooden block

Which of these physical properties do all four objects have in common?

Response:

They all attract the same metal objects and are not soluble in water.

Correct Response:

They all are not soluble in water and have the same physical state.

TEKS: 5.5A - classify matter based on measurable, testable, and observable physical properties, including mass, magnetism, physical state (solid, liquid, and gas), relative density (sinking and floating using water as a reference point), solubility in water, and the ability to conduct or insulate thermal energy or electric energy